

TITLE OF THE INVENTION
Semiconductor Encapsulating Epoxy Resin Composition
and Semiconductor Device

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This invention relates to an epoxy resin composition for semiconductor encapsulation which is effectively 10 moldable and cures into a product having solder crack resistance, flame retardance and moisture-proof reliability. It also relates to a semiconductor device encapsulated with a cured product of the composition.

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BACKGROUND OF THE INVENTION

The current mainstream in the semiconductor industry resides in diodes, transistors, ICs, LSIs and VLSIs of the resin encapsulation type. Epoxy resins have superior 20 moldability, adhesion, electrical properties, mechanical properties, and moisture resistance to other thermosetting resins. It is thus a common practice to encapsulate semiconductor devices with epoxy resin compositions. Semiconductor devices are now used in every area of the modern society, for example, in electric appliances and 25 computers. As a guard against accidental fire, the semiconductor encapsulating materials are required to be flame retardant.

Halogenated epoxy resins (typically chlorinated and brominated epoxy resins) combined with antimony trioxide are often blended in epoxy resin compositions in order to 30 enhance flame retardance. This combination of a halogenated epoxy resin with antimony trioxide has great radical-trapping and air-shielding effects in the vapor phase, thus conferring a high fire-retarding effect.

In a high-temperature environment, however, such flame retardants as halides (typically chlorides and bromides) and antimony compounds are decomposed to give rise to chemical 35

reaction at connections between gold wires and aluminum lines. This results in an increased resistance at the connections or even disconnection, inviting malfunction. In addition, the halogenated epoxy resins generate noxious gases during combustion, and antimony trioxide has powder toxicity. Given their negative impact on human health and the environment, it is desirable to entirely exclude these fire retardants from resin compositions.

In view of the above demand, studies have been conducted on the use of hydroxides such as $\text{Al}(\text{OH})_3$ and $\text{Mg}(\text{OH})_2$ or phosphorus-containing fire retardants in place of halogenated epoxy resins and antimony trioxide. Unfortunately, because of various problems associated with the use of these alternative compounds, such as inferior curability of the resin composition during molding and poor moisture resistance in the cured product, they are not yet ready for practical application.

SUMMARY OF THE INVENTION

An object of the invention is to provide an epoxy resin composition for semiconductor encapsulation which is free of harmful halogenated epoxy resins and antimony compounds, is effectively moldable and cures into a product having improved solder crack resistance, flame retardance and reliability. Another object is to provide a semiconductor device encapsulated with a cured product of the composition.

The invention provides a semiconductor encapsulating epoxy resin composition comprising (A) an epoxy resin, (B) a phenolic resin curing agent, (C) a molybdenum compound, (D) a silicon compound, and (E) an inorganic filler. The silicon compound (D) is selected from the group consisting of (D-i) an organopolysiloxane of the following average compositional formula (1):

$\text{R}^1_a\text{SiO}_{(4-a)/2}$ (1)

wherein R^1 is a substituted or unsubstituted monovalent hydrocarbon group, and "a" is a positive number of 0.8 to 3,

with the proviso that two R¹ groups, taken together, may form an alkylene group, (D-ii) a cured product of organopolysiloxane, and (D-iii) a block copolymer obtained by reacting an epoxy resin or an alkenyl group-bearing epoxy resin with an organopolysiloxane of the following average compositional formula (2):



wherein R² is a substituted or unsubstituted monovalent hydrocarbon group, m is a positive number of 0.001 to 0.2, n is a positive number of 1.8 to 2.1, and m+n is 1.801 to 2.3, the number of silicon atoms in a molecule is an integer of 10 to 1,000, and the number of hydrogen atoms directly attached to silicon atoms is 1 to 5, addition reaction taking place between epoxy groups on the epoxy resin or alkenyl groups on the alkenyl group-bearing epoxy resin and silicon-attached hydrogen atoms (i.e., SiH groups) on the organopolysiloxane.

The semiconductor encapsulating epoxy resin compositions of the invention are effectively molded and cure into products which have an excellent fire retardance, moisture resistance and solder cracking resistance despite the absence of halogenated epoxy resins and antimony compounds (e.g., antimony trioxide).

25 DETAILED DESCRIPTION OF THE INVENTION

Component (A) is an epoxy resin which is not critical as long as it has at least two epoxy groups per molecule. This epoxy resin is different from the block copolymer (D-iii) to be described later in that it does not contain a siloxane structure in the molecule. Illustrative examples of suitable epoxy resins include novolac-type epoxy resins such as phenolic novolac epoxy resins and cresol novolac epoxy resins, triphenolalkane epoxy resins such as triphenolmethane type epoxy resin and triphenolpropane type epoxy resin, phenolaralkyl epoxy resins, biphenyl skeleton-containing aralkyl epoxy resins, biphenyl epoxy resins, heterocyclic epoxy resins, naphthalene ring-containing epoxy

resins, bisphenol-type epoxy resins such as bisphenol A epoxy compounds and bisphenol F epoxy compounds, and stilbene epoxy resins. Any one or combination of two or more of these epoxy resins may be employed. Halogenated
5 epoxy resins are excluded.

No particular limit is imposed on the phenolic resin serving as curing agent (B) in the invention, so long as the phenolic resin has at least two phenolic hydroxy groups in a molecule. Illustrative examples of typical phenolic resin
10 curing agents include novolac-type phenolic resins such as phenolic novolac resins and cresol novolac resins, naphthalene ring-containing phenolic resins, triphenolalkane phenolic resins such as triphenolmethane type phenolic resin and triphenolpropane type phenolic resin, phenolaralkyl
15 phenolic resins, biphenyl skeleton-containing aralkyl phenolic resins, biphenyl phenolic resins, alicyclic phenolic resins, heterocyclic phenolic resins, and bisphenol-type phenolic resins such as bisphenol A and bisphenol F. Any one or combination of two or more of these
20 phenolic resins may be employed.

The relative proportions of the epoxy resin (A) and the phenolic resin curing agent (B) used in the epoxy resin compositions are not subject to any particular limits, although it is preferred that the amount of phenolic
25 hydroxyl groups in the curing agent (B) be from 0.5 to 1.5 moles, and especially 0.8 to 1.2 moles, per mole of epoxy groups in the epoxy resin (A).

The epoxy resin composition of the invention contains a molybdenum compound as the flame retardant (C). Exemplary
30 molybdenum compounds are molybdenum oxides, molybdenum borides, molybdenum silicides, molybdenum esters, and molybdic salts such as molybdenum boride, molybdenum disilicide, molybdenum acetylacetone, molybdenum (IV) oxide, molybdenum (V) oxide, molybdenum (VI) oxide, zinc
35 molybdate, calcium molybdate carbonate, and calcium molybdate. The molybdenum compound by itself is known to have a smoke-reducing and charring effect in burning

plastic. Although like antimony trioxide, molybdenum compounds are conventionally used in combination with halogenated resins, it has been found by the inventors that flame retardance is exerted by combining molybdenum 5 compounds with the organopolysiloxane, organopolysiloxane cured product or block copolymer to be described later. Since the molybdenum compounds are free from powder toxicity as found with antimony trioxide, they are quite safe flame retardants. Of these, zinc molybdate is especially 10 preferred since it does not affect the curability of the epoxy resin.

To achieve a satisfactory flame retardant effect, zinc molybdate must be uniformly dispersed in the epoxy resin composition. To improve the dispersibility, zinc molybdate 15 is preferably supported on an inorganic carrier such as silica or talc. Suitable inorganic carriers for supporting zinc molybdate include silicas such as fused silica and crystalline silica, talc, alumina, silicon nitride, aluminum nitride, boron nitride, titanium oxide, and glass fibers. 20 The zinc molybdate-carrying powder should preferably have a mean particle diameter of 0.1 to 40 μm , more preferably 0.2 to 15 μm , and most preferably 0.5 to 5 μm and a specific surface area of 0.5 to 50 m^2/g , and more preferably 0.7 to 10 m^2/g as measured by the BET method.

25 It is noted that the mean particle diameter can be determined as the weight average value (or median diameter) based on the laser light diffraction technique, for example.

In the flame retardant comprising zinc molybdate supported on the inorganic carrier, the content of zinc 30 molybdate is preferably 5 to 40% by weight and more preferably 10 to 30% by weight. Less contents of zinc molybdate may fail to provide satisfactory flame retardance whereas excessive contents may detract from flow during molding and curability.

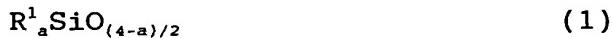
35 The zinc molybdate on inorganic carrier is commercially available under the trade name of KEMGARD

series, such as KEMGARD 1260, 1261, 1270, 1271 and 911C from Sherwin-Williams Co.

An appropriate amount of the flame retardant (C) in the form of zinc molybdate on inorganic carrier is 1 to 120 parts, more preferably 3 to 100 parts, and especially 5 to 100 parts by weight per 100 parts by weight of the epoxy resin (A) and the phenolic resin curing agent (B) combined. The amount of zinc molybdate alone in the flame retardant (when the molybdenum compound is blended without supporting it on an inorganic carrier, the amount of the molybdenum compound itself) is preferably 0.05 to 35 parts, more preferably 0.1 to 30 parts, and especially 0.2 to 20 parts by weight per 100 parts by weight of the epoxy resin (A) and the phenolic resin curing agent (B) combined. Less amounts may fail to provide satisfactory flame retardance whereas excessive amounts may detract from the flow and curability of the composition.

The composition of the invention contains as component (D) at least one silicon compound selected from among:

(D-i) an organopolysiloxane of the following average compositional formula (1):



wherein R^1 is a substituted or unsubstituted monovalent hydrocarbon group, and "a" is a positive number of 0.8 to 3, with the proviso that two R^1 groups, taken together, may form an alkylene group,

(D-ii) a cured product of organopolysiloxane, and

(D-iii) a block copolymer obtained by reacting an epoxy resin or an alkenyl group-bearing epoxy resin with an organopolysiloxane of the following average compositional formula (2):



wherein R^2 is a substituted or unsubstituted monovalent hydrocarbon group, m is a positive number of 0.001 to 0.2, n is a positive number of 1.8 to 2.1, and $m+n$ is 1.801 to 2.3, the number of silicon atoms in a molecule is an integer of 10 to 1,000, and the number of hydrogen atoms directly

attached to silicon atoms is 1 to 5, addition reaction taking place between epoxy groups on the epoxy resin or alkenyl groups on the alkenyl group-bearing epoxy resin and silicon-attached hydrogen atoms (i.e., SiH groups) on the 5 organopolysiloxane.

Component (D-i) is an organopolysiloxane of the average compositional formula (1): $R^1_aSiO_{(4-a)/2}$ which is typically a non-crosslinkable silicone fluid (oil) or silicone resin (hydrolysis condensate) whereas component (D-10 ii) is a cured product of organopolysiloxane which is typically a crosslinked silicone rubber or silicone resin. It is known that the addition of a silicone polymer to a plastic material is effective for improving flame retardance because the silicon polymer forms a flame retardant silicon 15 carbide (Si-C) coating upon combustion. According to the invention, component (D-i) and/or (D-ii) is used in combination with the molybdenum compound, especially zinc molybdate, whereby the formation of a silicon carbide (Si-C) coating is promoted to enhance flame retardance.

In formula (1), O represents an oxygen atom forming a siloxane structure ($=Si-O-Si=$). R^1 is a substituted or unsubstituted monovalent hydrocarbon group attached to a silicon atom forming the siloxane structure. The unsubstituted monovalent hydrocarbon groups represented by 25 R^1 are preferably those having 1 to 12 carbon atoms, and more preferably 1 to 8 carbon atoms, for example, alkyl groups such as methyl, ethyl, propyl, isopropyl, butyl, isobutyl, tert-butyl, pentyl, hexyl, cyclohexyl, and octyl; alkenyl groups such as vinyl, allyl, propenyl, butenyl, and 30 hexenyl; aryl groups such as phenyl, tolyl and naphthyl; and aralkyl groups such as benzyl, phenylethyl, and phenylpropyl. The substituted monovalent hydrocarbon groups include those corresponding to the foregoing unsubstituted monovalent hydrocarbon groups in which some or all hydrogen 35 atoms are replaced by fluorine atoms, cyano, hydroxyl, alkoxy, amino or mercapto groups as well as monovalent hydrocarbon groups containing imino groups, epoxy groups,

carboxyl groups, carbinol groups, (methyl)styryl groups, (meth)acrylic groups, polyether groups, higher fatty acid groups, higher fatty acid ester groups, and long-chain alkyl groups of at least 13 carbon atoms. Two R¹ groups, taken 5 together, may form an alkylene group of about 1 to 8 carbon atoms, especially about 2 to 6 carbon atoms, such as methylene, ethylene, trimethylene, tetramethylene, methylethylene, or hexamethylene.

Letter "a" is a positive number of 0.8 to 3, 10 preferably 1 to 2.7. The organopolysiloxane may be a linear or cyclic one composed mainly of R¹₂SiO_{2/2} units, or a copolymer of three-dimensional network (or resinous) structure containing essentially R¹SiO_{2/3} units and/or SiO_{4/2} units and having R¹₃SiO_{1/2} units or R¹₂SiO_{2/2} units optionally 15 combined therewith.

The organopolysiloxane (D-i) represented by formula (1) preferably has a number average degree of polymerization of 2.5 to 1,000, and more preferably 3 to 200. It is noted 20 that the degree of polymerization corresponds to the number of silicon atoms in the molecule. An organopolysiloxane with a number average degree of polymerization of less than 2.5 may give rise to problems of volatility and compatibility because of a low molecular weight whereas an organopolysiloxane with a number average degree of 25 polymerization of more than 1,000 may become high viscous and less dispersible.

Among component (D-i), polydimethylsiloxane and polymethylphenylsiloxane are typical of the non-functional linear organopolysiloxane which does not contain a 30 functional group-substituted monovalent hydrocarbon group in the molecule. Modified organopolysiloxanes include amino-modified polyorganosiloxanes, epoxy-modified polyorganosiloxanes, carboxyl-modified polyorganosiloxanes, carbinol-modified polyorganosiloxanes, methacryl-modified 35 polyorganosiloxanes, mercapto-modified polyorganosiloxanes, phenol-modified polyorganosiloxanes, polyorganosiloxanes modified to be reactive at one end, polyorganosiloxanes

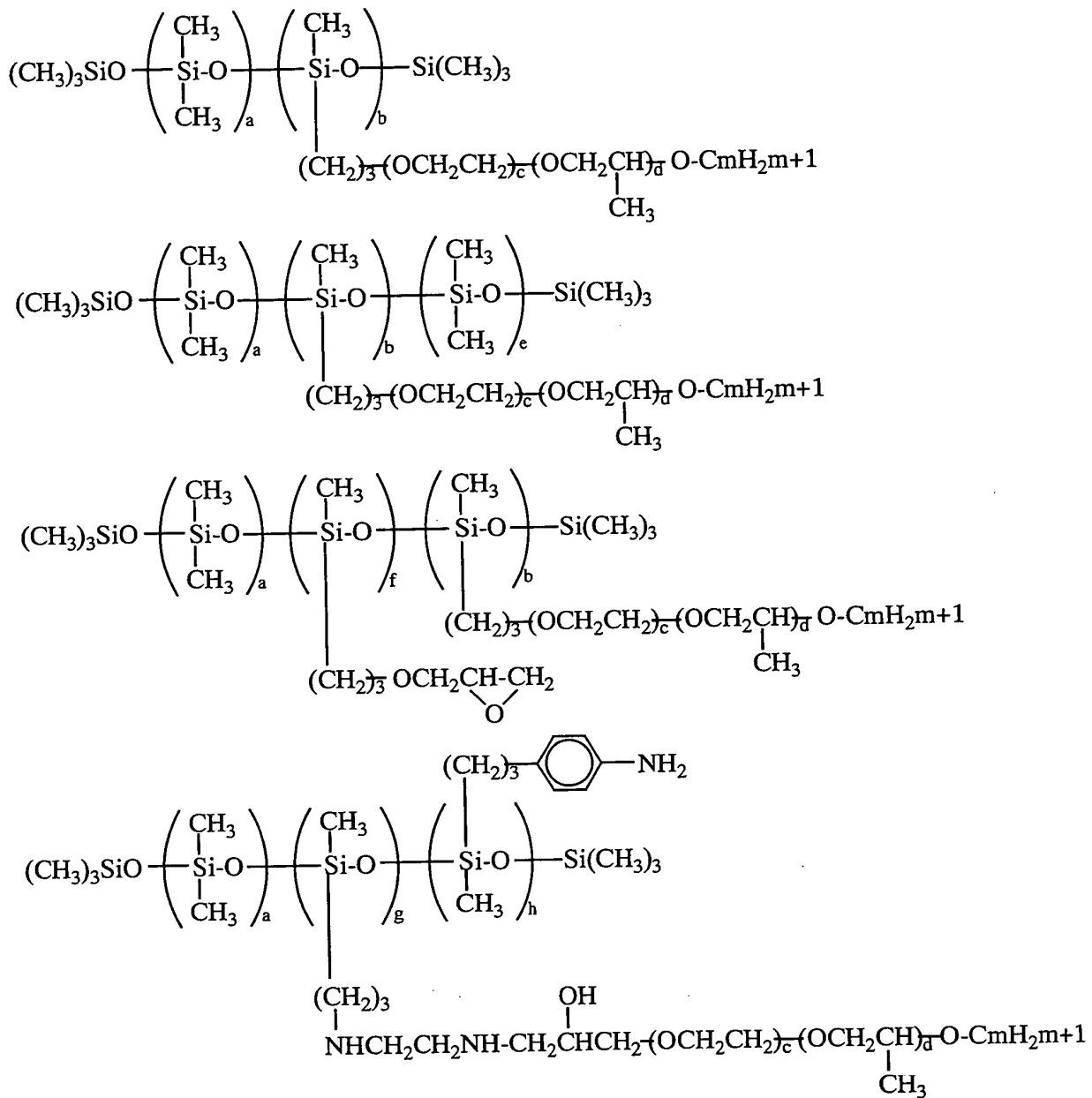
modified with different functional groups, polyether-modified polyorganosiloxanes, methylstyryl-modified polyorganosiloxanes, long-chain (usually about C₁₃₋₃₀) alkyl-modified polyorganosiloxanes, higher fatty acid ester-modified polyorganosiloxanes, higher fatty acid-containing modified polyorganosiloxanes, and fluorine-modified polyorganosiloxanes such as trifluoropropyl-methylpolysiloxane. These organopolysiloxanes may be linear, branched or cyclic.

Polyorganosiloxane resins are generally divided into polymethylsiloxane resins and poly(methyl)phenylsiloxane resins. Those having a carbon-to-carbon double bond in the structure are known to be highly flame retardant. The polymethylsiloxane resins are generally copolymers containing SiO₂ and/or CH₃SiO_{3/2} essentially and having (CH₃)₂SiO, (CH₂=CH)(CH₃)₂SiO_{1/2} and (CH₃)₃SiO_{1/2} structural units combined therewith. The poly(methyl)phenylsiloxane resins are generally copolymers containing C₆H₅SiO_{3/2} and/or SiO₂ essentially and having (C₆H₅)(CH₃)₂SiO_{1/2}, (CH₃)₂SiO, CH₃SiO_{3/2}, (CH₃)₃SiO_{1/2}, (CH₂=CH)(CH₃)₂SiO_{1/2}, C₆H₅(CH₃)SiO, and (C₆H₅)₂SiO structural units combined therewith, and are more heat resistant than the polymethylsiloxane resins.

The cured product of organopolysiloxane used as component (D-ii) should have a crosslinked structure and is selected from the silicone rubbers and silicone resins shown below. Included are organopolysiloxane rubbers cured by addition reaction in the presence of a platinum catalyst, organopolysiloxane rubbers cured by condensation reaction (the condensation reaction including reactions of eliminating water, hydrogen, alcohol, oxime, amine, amide, carboxylic acid, and ketone), organopolysiloxane rubbers heat cured with organic peroxides, organopolysiloxane rubbers cured by irradiating ultraviolet radiation, and organopolysiloxane resins cured by any of the foregoing reactions. Organopolysiloxane cured products having a functional group such as a silanol, hydroxy, carboxy, vinyl, amino, mercapto, epoxy, methoxy, or ethoxy group in the

structure may also be used. These organopolysiloxane cured products are ground by means of a suitable pulverizer and the resulting powder is used. Since the advanced mold for molding electronic parts such as ICs has a reduced gate size
5 matching with the desired size reduction of packages, particles of the organopolysiloxane cured product should preferably have a mean particle size of up to 50 μm , and especially up to 20 μm from the molding standpoint. The lower limit is usually 0.01 μm , and preferably 0.1 μm .

10 Of the above-described organopolysiloxanes (D-i), polyether-modified polyorganosiloxane compounds having polyether groups with a high affinity to the matrix resin are preferred from the standpoints of marking on the package surface and adhesion. The polyether-modified
15 polyorganosiloxane compounds are exemplified by the following compounds.



In the formulae, letters a to h and m have the following meaning.

- a: integer of 5 to 200
- b: integer of 1 to 5
- c: integer of 0 to 30
- d: integer of 0 to 30
- c+d: integer of 10 to 60
- e: integer of 1 to 50

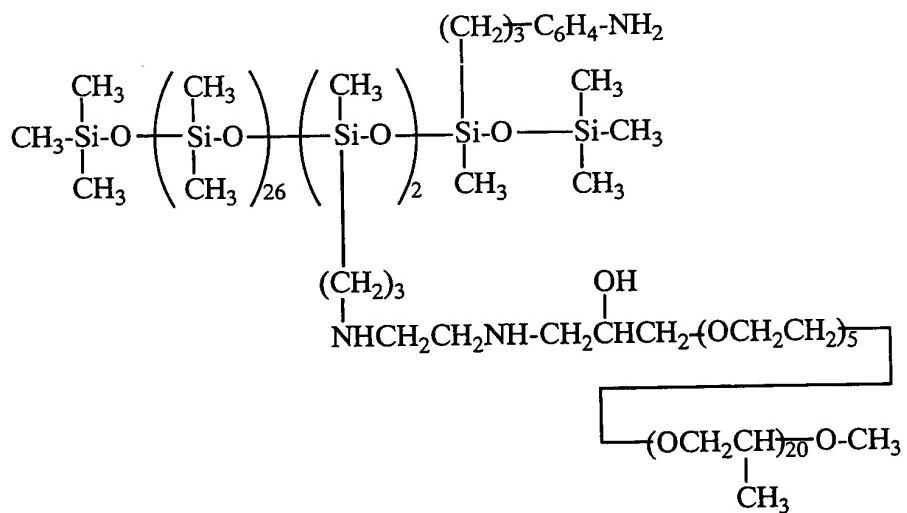
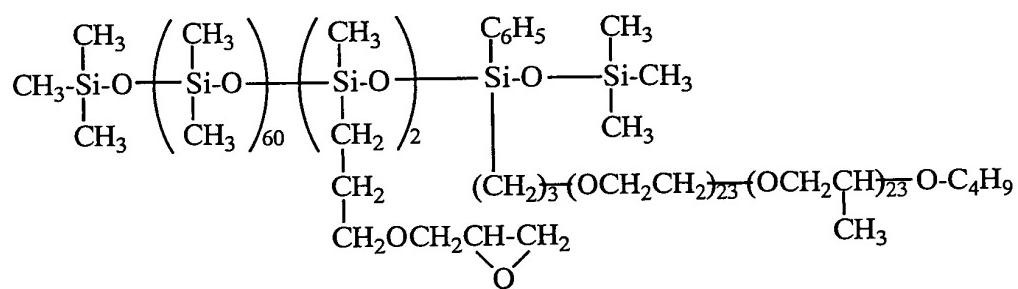
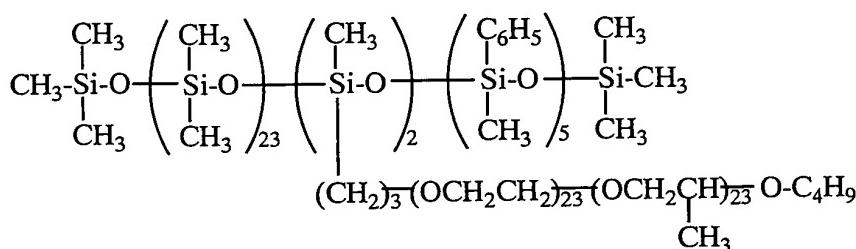
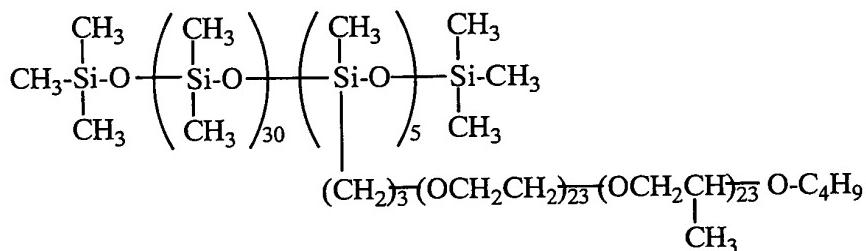
f: integer of 1 to 5

q: integer of 1 to 5

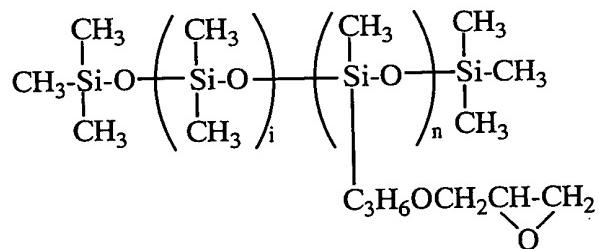
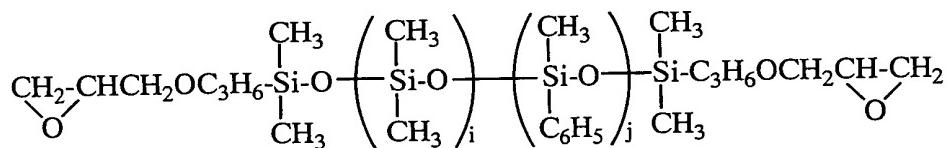
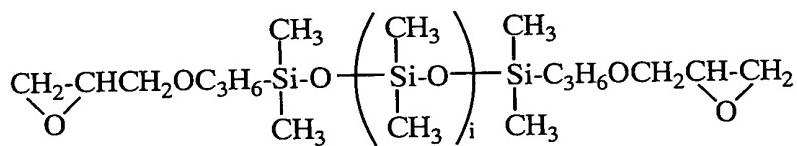
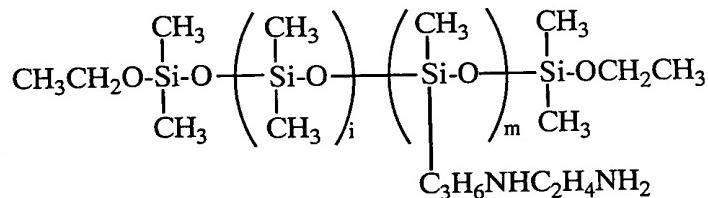
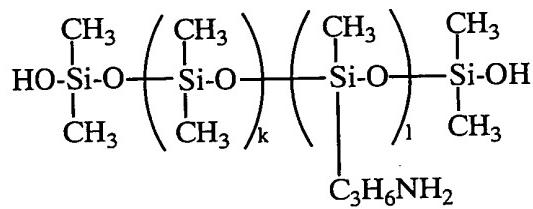
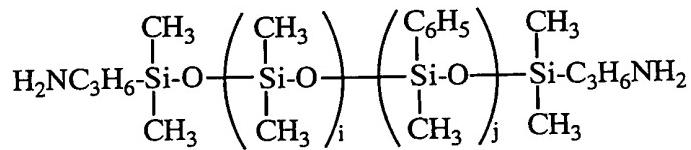
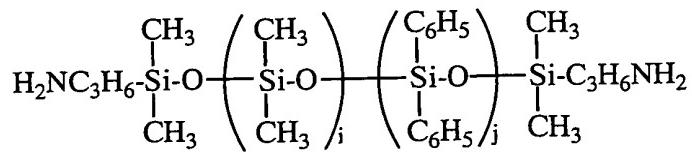
h: integer of 0 to 5

m: integer of 1 to 6

5 Illustrative examples of the polyether-modified organopolysiloxane are given below.



It is acceptable to use the polyether-modified organopolysiloxane as a compatibilizing agent in combination with another organopolysiloxane compound such as an amino-
5 or epoxy-modified organopolysiloxane having an amino functional or epoxy functional group attached to a silicon atom at an end or intermediate of the molecular chain as exemplified below or an organopolysiloxane cured product as mentioned above.



In the formulae, letters i to n have the following meaning.

i: integer of 5 to 200

j: integer of 1 to 50

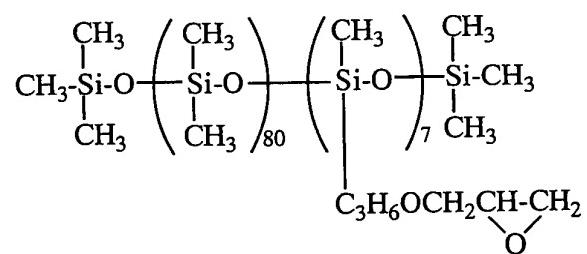
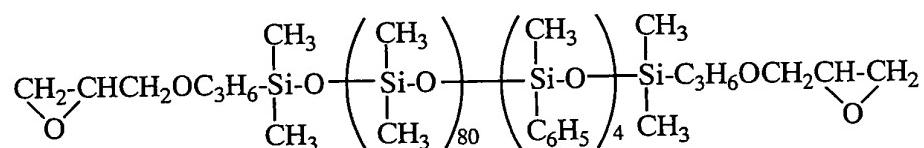
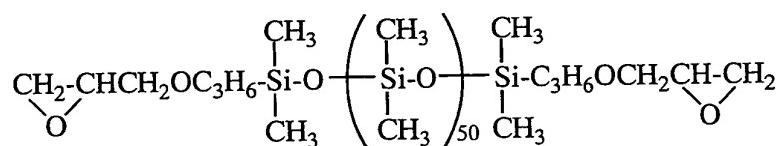
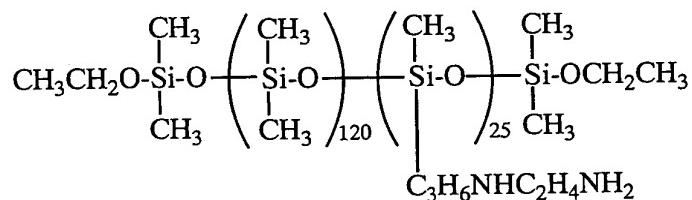
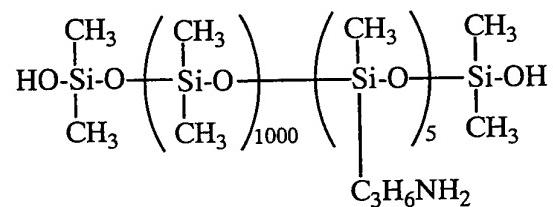
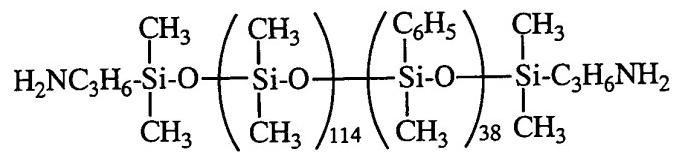
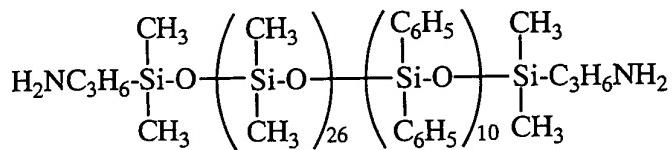
k: integer of 50 to 2000

l: integer of 1 to 10

m: integer of 1 to 10

n: integer of 1 to 10

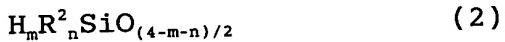
5 Illustrative examples of the amino- and epoxy-modified
organopolysiloxanes are given below.



An appropriate amount of component (D-i) or (D-ii) blended is 0.1 to 8 parts, and especially 0.2 to 4 parts by weight per 100 parts by weight of the epoxy resin (A) and

phenolic resin curing agent (B) combined. Less than 0.1 part may fail to achieve a satisfactory flame retardance whereas more than 8 parts may lead to the disadvantages of an increased viscosity and a reduced mechanical strength.

5 Component (D-iii) is a block copolymer obtained by reacting an epoxy resin or an alkenyl group-bearing epoxy resin with an organopolysiloxane of the following average compositional formula (2):



10 wherein R^2 is a substituted or unsubstituted monovalent hydrocarbon group, m is a positive number of 0.001 to 0.2, n is a positive number of 1.8 to 2.1, and $m+n$ is 1.801 to 2.3, the number of silicon atoms in a molecule is an integer of 10 to 1,000, and the number of hydrogen atoms directly attached to silicon atoms is 1 to 5, preferably 1 to 3, and most preferably equal to 2, addition reaction taking place between epoxy groups on the epoxy resin or alkenyl groups on the alkenyl group-bearing epoxy resin and silicon-attached hydrogen atoms (i.e., SiH groups) on the organopolysiloxane.

15 As previously described, it is known that the addition of a silicone polymer to a plastic material is effective for improving flame retardance because the silicon polymer forms a flame retardant silicon carbide (Si-C) coating upon combustion. According to the invention, component (D-iii)

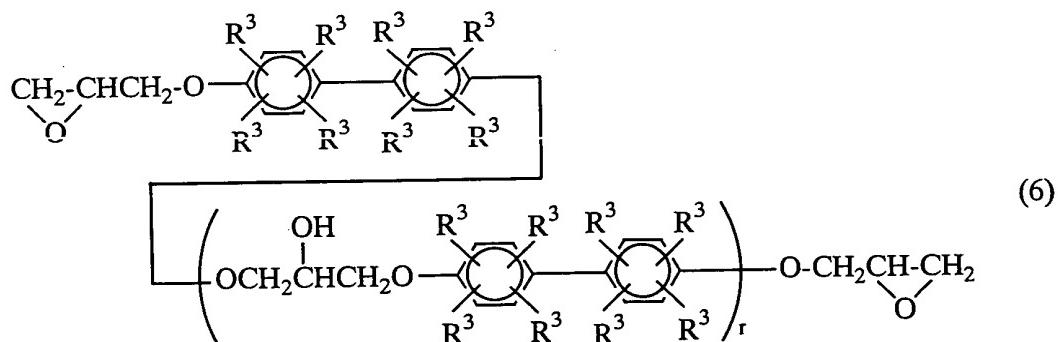
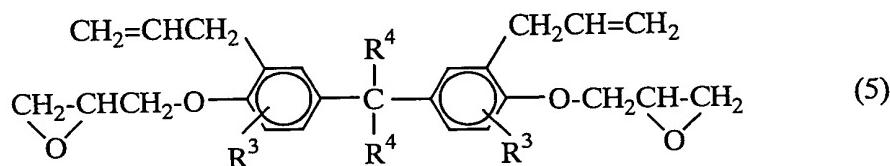
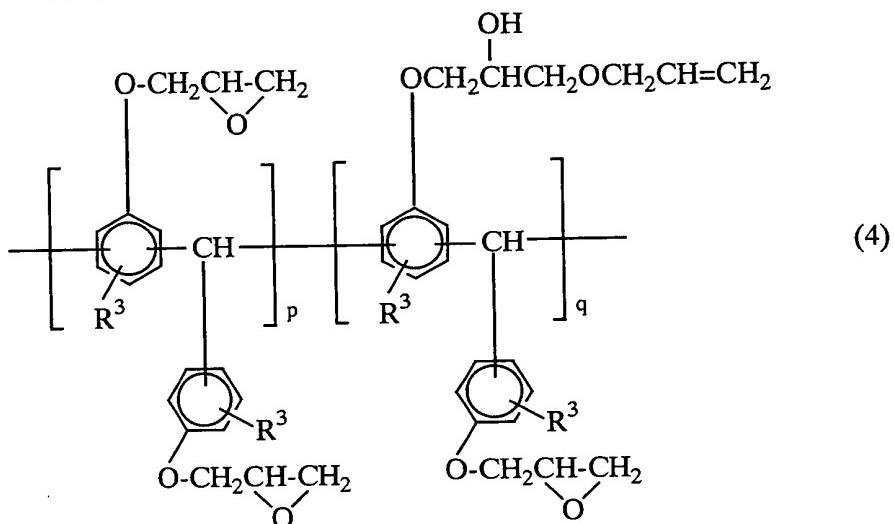
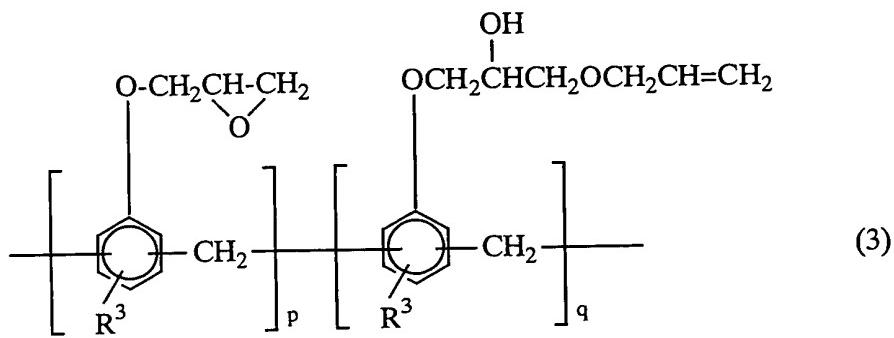
20 is used in combination with the molybdenum compound, especially zinc molybdate, whereby the formation of a silicon carbide (Si-C) coating is promoted to enhance flame retardance. In particular, since the block copolymer (D-iii) has an epoxy resin structure as well as an

25 organopolysiloxane structure in the molecule, it can be microscopically dispersed in a resin matrix of the epoxy resin and the phenolic resin curing agent to form an island-in-sea structure, which contributes to a further improvement in flame retardance.

30 The epoxy resin used as one component of the copolymer may be selected from the same epoxy resins exemplified for component (A), especially difunctional epoxy resins having

two epoxy groups per molecule, for example, difunctional epoxy resins represented by the following formula (6). The alkenyl group-bearing epoxy resin used herein may be selected from a variety of epoxy resins which have at least one, and preferably at least two epoxy groups such as glycidyloxy groups attached to an aromatic ring and at least one, and preferably at least two alkenyl groups, for example, alkenyl group-bearing epoxy resins of the structure represented by the following formulae (3) to (5).

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Herein, R^3 is hydrogen, methyl or bromine, R^4 is hydrogen, methyl or trifluoromethyl, p and q are natural numbers, and preferably $p+q$ is from 2 to 50, more preferably from 2 to 20, and r is an integer of 0 to 5.

In the average compositional formula (2) representing the organohydrogenpolysiloxane, the substituted or unsubstituted monovalent hydrocarbon groups attached to silicon atoms represented by R² are preferably those of 1 to 5 10 carbon atoms, and especially those free of aliphatic unsaturation, for example, alkyl groups such as methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, pentyl, neopentyl, hexyl, cyclohexyl, octyl, and decyl; alkenyl groups such as vinyl, allyl, propenyl, and butenyl; 10 aryl groups such as phenyl and tolyl; and aralkyl groups such as benzyl and phenylethyl; substituted ones of the foregoing hydrocarbon groups in which some or all of the hydrogen atoms are replaced by fluorine atoms or alkoxy groups, for example, fluorine-substituted monovalent 15 hydrocarbon groups such as fluorine-substituted alkyl groups, e.g., 3,3,3-trifluoropropyl and alkoxy-substituted monovalent hydrocarbon groups such as alkoxy-substituted alkyl groups, e.g., methoxymethyl, methoxyethyl, ethoxymethyl and ethoxyethyl. Of these, the unsubstituted 20 monovalent hydrocarbon group not having an aliphatic unsaturated bond, such as an alkyl group, aryl group and aralkyl group is preferred, and especially methyl and phenyl are most preferred.

In formula (2), m and n are as defined above, 25 preferably 0.005 ≤ m ≤ 0.1, 1.9 ≤ n ≤ 2.1, and 1.905 ≤ m+n ≤ 2.2, and more preferably 0.01 ≤ m ≤ 0.1, 1.95 ≤ n ≤ 2.0, and 1.96 ≤ m+n ≤ 2.1. The number of silicon atoms in a molecule is an integer of about 10 to 1,000, preferably about 20 to 400 and more preferably about 40 to 200.

Exemplary organohydrogenpolysiloxanes include both end 30 trimethylsiloxy group-blocked methylhydrogenpolysiloxane, both end trimethylsiloxy group-blocked dimethylsiloxane-methylhydrogensiloxane copolymers, both end dimethylhydrogensiloxy group-blocked dimethylpolysiloxane, 35 both end dimethylhydrogensiloxy group-blocked dimethylsiloxane-methylhydrogensiloxane copolymers, both end dimethylsiloxane-methylhydrogensiloxane copolymers, both end

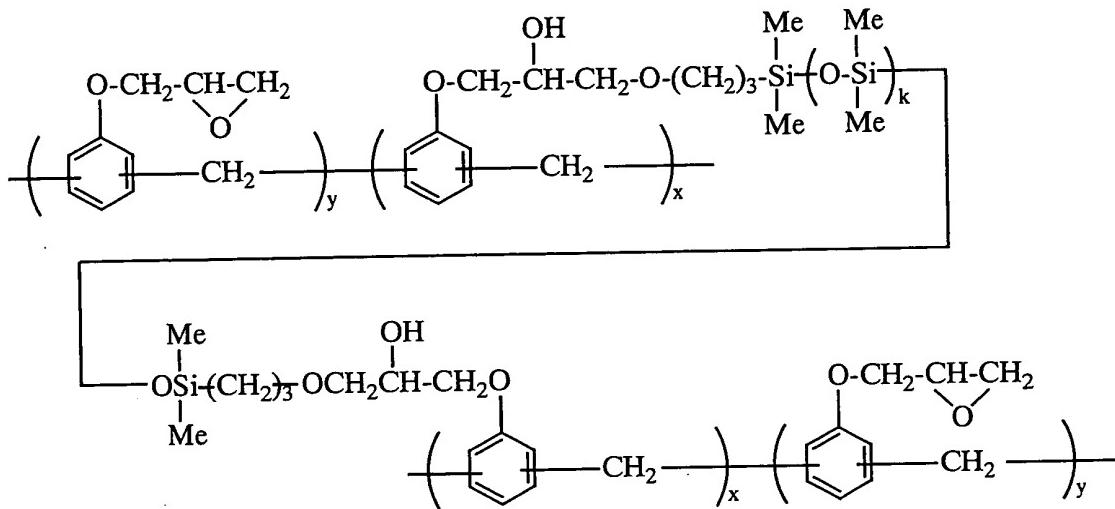
dimethylhydrogensiloxy group-blocked dimethylsiloxane-diphenylsiloxane copolymers, and both end dimethylhydrogensiloxy group-blocked dimethylsiloxane-methylphenylsiloxane copolymers. These are basically of a linear structure although they may partially contain a branched siloxane structure.

The organopolysiloxane desirably has a molecular weight of 700 to 75,000 though not limited thereto. When a copolymer of an organopolysiloxane with a molecular weight of 700 to 75,000 is blended in an epoxy resin composition, the copolymer is not compatible with the matrix and forms a microscopic island-in-sea structure. With a molecular weight of less than 700, the copolymer may become compatible with the matrix with the island-in-sea structure lost. With a molecular weight of more than 75,000, the copolymer may form a large island-in-sea structure. In either case, cured products may lose flame retardance and low stress properties.

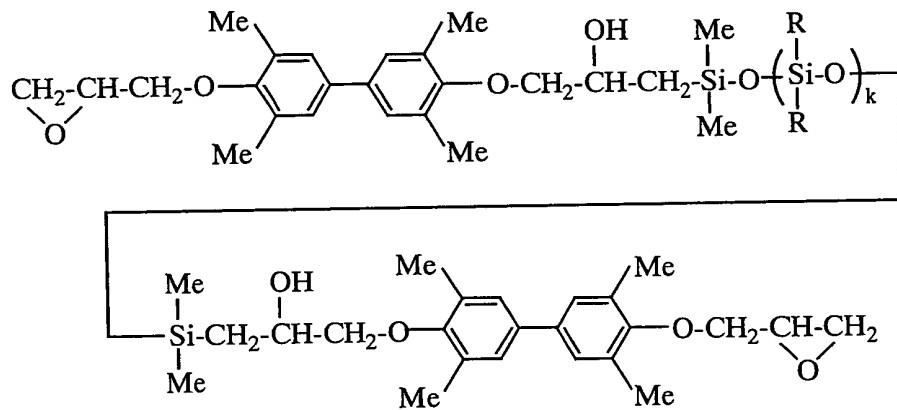
The copolymer of the alkenyl group-bearing epoxy resin with the organohydrogenpolysiloxane can be prepared by effecting addition reaction known as hydrosilylation reaction between alkenyl groups on the epoxy resin and silicon atom-attached hydrogen atoms (i.e., SiH groups) on the organohydrogenpolysiloxane. The copolymer of the alkenyl group-free epoxy resin (e.g., difunctional epoxy resin) with the organohydrogenpolysiloxane can be prepared by effecting direct addition reaction between epoxy groups on the epoxy resin and SiH groups on the organohydrogenpolysiloxane.

As the block copolymer, use may be made of the well-known copolymers described in USP 4,902,732 (JP-B 61-48544), USP 4,877,822, and USP 5,053,445 (JP-B 63-60069).

Illustrative examples of these copolymers are given below.



Herein, Me is methyl, x and y are positive integers satisfying $10 \leq y/x \leq 30$, and k is an integer of 10 to 500.



5

Herein, Me is methyl, R is C₁₋₁₀ alkyl or phenyl, and k is an integer of 10 to 500.

An appropriate amount of component (D-iii) blended is 1 to 10 parts, and especially 2 to 8 parts by weight, calculated as organopolysiloxane, per 100 parts by weight of the epoxy resin (A) and phenolic resin curing agent (B) combined. Less than 1 part may fail to achieve a satisfactory flame retardance whereas more than 10 parts may invite an increased viscosity and detract from the reliability, especially solder crack resistance of the semiconductor device encapsulated therewith.

The inorganic filler (E) included in the epoxy resin compositions of the invention may be any suitable inorganic filler commonly used in epoxy resin compositions.

Illustrative examples include silicas such as fused silica
5 and crystalline silica, alumina, silicon nitride, aluminum
nitride, boron nitride, titanium oxide, and glass fibers.
No particular limit is imposed on the mean particle diameter
and shape of these inorganic fillers, although the use of
spherical fused silica having a mean particle diameter of 1
10 to 40 μm , and especially 2 to 25 μm is preferred because it
endows the epoxy resin composition with good molding and
flow characteristics.

The inventive epoxy resin compositions have inorganic
filler loadings of preferably 300 to 1,000 parts by weight,
15 and more preferably 500 to 900 parts by weight per 100 parts
by weight of the mixture of the epoxy resin (A), the
phenolic resin curing agent (B) and the optional block
copolymer (D-iii). At less than 300 parts by weight, the
epoxy resin composition would have a large coefficient of
20 expansion, resulting in greater stress on the semiconductor
device and a decline in the device characteristics.

Moreover, the proportion of resin relative to the overall
composition becomes larger, sometimes failing to attain the
fire retardance that is the object of this invention. On
25 the other hand, more than 1,000 parts by weight of the
inorganic filler would result in an excessive rise in
viscosity during molding and thus impede molding. The
content of inorganic filler within the epoxy resin
composition (excluding the flame retardant (C)) is
30 preferably 70 to 92% by weight, and especially 80 to 90% by
weight.

The inorganic filler used herein is preferably surface
treated beforehand with a coupling agent such as a silane
coupling agent or a titanate coupling agent in order to
35 increase the bonding strength between the resin and the
inorganic filler. Preferred examples of such coupling
agents include epoxy silanes such as γ -glycidoxypropyl-

trimethoxysilane, γ -glycidoxypromethylmethyldiethoxysilane, and β -(3,4-epoxycyclohexyl)ethyltrimethoxysilane; aminosilanes such as N- β -(aminoethyl)- γ -aminopropyltrimethoxysilane, γ -aminopropyltriethoxysilane, and N-phenyl- γ -amino-

- 5 propyltrimethoxysilane; and mercaptosilanes such as γ -mercaptopropyltrimethoxysilane and γ -mercaptopropylmethyldimethoxysilane. No particular limitation is imposed on the amount of coupling agent used for surface treatment or the method of surface treatment.

10 In the practice of this invention, use is preferably made of a curing accelerator to promote the curing reaction between the epoxy resin and the curing agent. The curing accelerator may be any suitable substance that promotes the curing reaction. Illustrative, non-limiting examples of
15 curing accelerators that may be used include organic phosphorus compounds such as triphenylphosphine, tributylphosphine, tri(p-methylphenyl)phosphine, tri(nonylphenyl)-phosphine, triphenylphosphine triphenylborane, and tetraphenylphosphine tetraphenylborate; tertiary amine
20 compounds such as triethylamine, benzylidemethylamine, α -methylbenzylidemethylamine, and 1,8-diazabicyclo[5.4.0]-undecene-7; and imidazole compounds such as 2-methylimidazole, 4-methylimidazole, 2-phenylimidazole, 2-phenyl-4-methylimidazole, 2-phenyl-4-hydroxymethylimidazole, 2-phenyl-4-methyl-5-hydroxymethylimidazole, and 2-phenyl-4,5-dihydroxymethylimidazole.

25 The semiconductor encapsulating epoxy resin compositions of the invention may also include various additives, if necessary. Illustrative examples include stress-lowering additives such as thermoplastic resins, thermoplastic elastomers, and synthetic organic rubbers; waxes such as carnauba wax, higher fatty acids, and synthetic waxes; colorants such as carbon black; and halogen trapping agents. Also, any of well-known diluents such as
30 n-butyl glycidyl ether, phenyl glycidyl ether, styrene
35

oxide, t-butyl phenyl glycidyl ether, dicyclopentadiene epoxide, phenol, cresol, and t-butylphenyl may be added for the purpose of lowering the viscosity.

The inventive epoxy resin compositions may be prepared by compounding the epoxy resin, curing agent, inorganic filler, and other components in predetermined proportions, thoroughly mixing these components together in a mixer or other appropriate apparatus, then melting and working the resulting mixture using a hot roll mill, kneader, extruder or the like. The worked mixture is then cooled and solidified, and subsequently ground to a suitable size so as to give a molding material.

The resulting epoxy resin compositions of the invention can be effectively used for encapsulating various types of semiconductor devices. The method of encapsulation most commonly used is low-pressure transfer molding. The epoxy resin composition of the invention is preferably molded at a temperature of about 150 to 180°C for a period of about 30 to 180 seconds, followed by postcuring at about 150 to 180°C for about 2 to 16 hours.

The semiconductor encapsulating epoxy resin compositions of the invention cure into products which have an excellent fire retardance, moisture resistance and solder cracking resistance. Owing to the absence of halogenated epoxy resins and antimony compounds (e.g., antimony trioxide), the epoxy resin compositions have no adverse impact on human health or the environment.

EXAMPLE

Examples of the invention and comparative examples are given below by way of illustration, and are not intended to limit the invention. All parts are by weight. Me stands for methyl.

The raw materials used in Examples are shown below.

Epoxy resins:

- (a) o-cresol novolac epoxy resin: EOCN1020-55 by Nippon Kayaku K.K. (epoxy equivalent 200)

(b) biphenyl epoxy resin: YX4000HK by Yuka Shell K.K.
(epoxy equivalent 190)

Curing agents:

- 5 (c) phenolic novolac resin: DL-92 by Meiwā Kasei K.K.
(phenolic equivalent 110)
(d) phenolic aralkyl resin: MEH-7800SS by Meiwā Kasei
K.K. (phenolic equivalent 175)

Zinc molybdate on inorganic carrier (zinc molybdate content 18 wt%):

- 10 KEMGARD 1261 (carrier: spherical silica with a mean particle size 0.5 μm and a specific surface area 5.5 m^2/g) by Sherwin-Williams Co.

KEMGARD 911C (carrier: talc with a mean particle size 2.0 μm and a specific surface area 2.0 m^2/g) by

15 Sherwin-Williams Co.

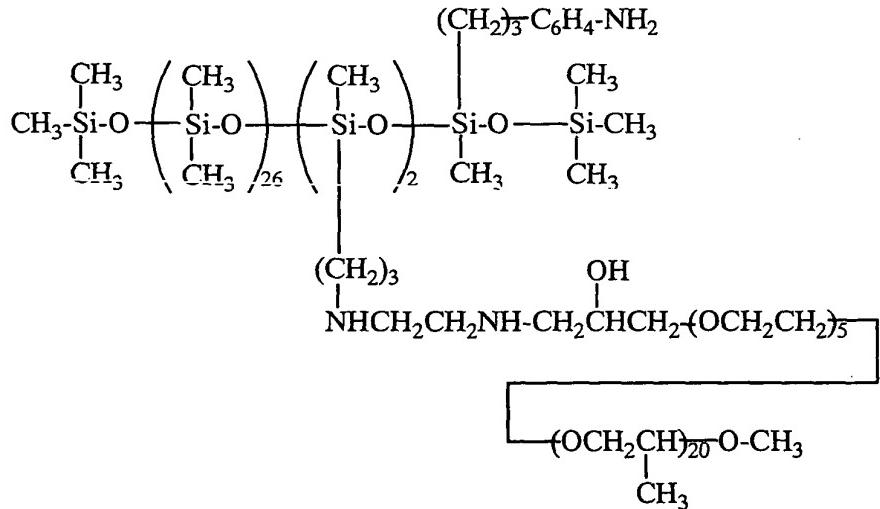
Inorganic filler:

Spherical fused silica (mean particle size 20 μm) by
Tatsumori K.K.

Polyorganosiloxane A:

- 20 KF96 (dimethylsilicone fluid (oil), viscosity 1,000 cs
at 25° C) by Shin-Etsu Chemical Co., Ltd.

Polyorganosiloxane B (polyether-modified silicone oil):



Polyorganosiloxane C:

KF54 (methylphenylsilicone fluid (oil), viscosity 400 cs at 25°C), by Shin-Etsu Chemical Co., Ltd.

Polyorganosiloxane D:

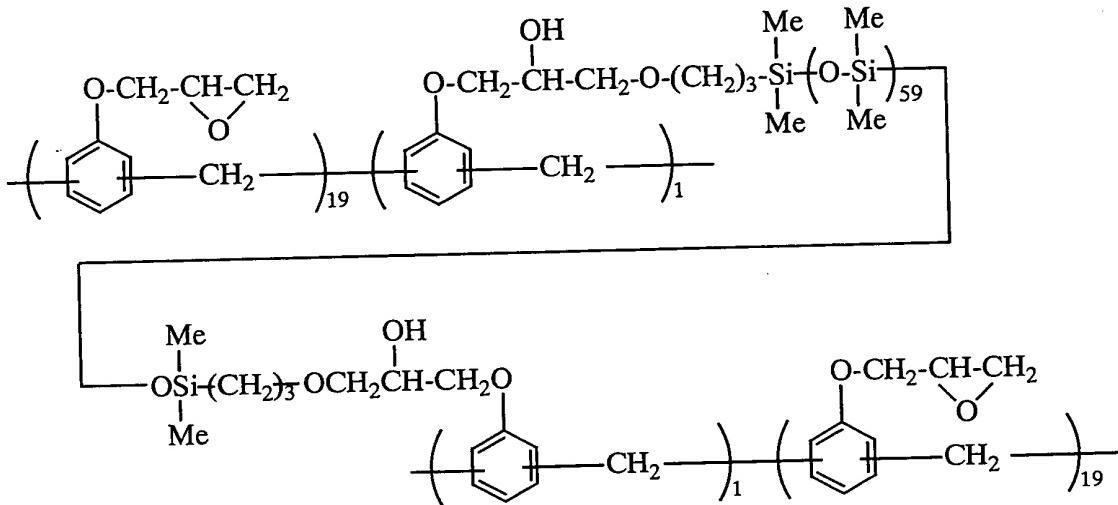
5 A liquid silicone resin of three-dimensional network structure was prepared by the following procedure.

10 A 1-liter flask equipped with a stirrer, condenser and thermometer was charged with 211 g (1 mol) of phenyltrichlorosilane and 143 g of toluene, and heated to an internal temperature of 40°C in an oil bath. A dropping funnel was charged with 64 g (2 mol) of methanol, which was added dropwise to the flask, with stirring, over one hour. Alkoxylation 15 reaction proceeded while hydrogen chloride gas given off during the reaction was removed out of the system. At the end of dropwise addition, agitation was continued for one hour at the internal temperature of 40°C for ripening. Next, the dropping funnel was 20 charged with 12 g (0.7 mol) of water, which was added dropwise to the flask, with stirring, over one hour. Hydrolytic condensation reaction proceeded while hydrogen chloride gas given off during the reaction was removed out of the system. At the end of dropwise addition, agitation was continued for one hour at the 25 internal temperature of 40°C for ripening. The toluene, the excess of methanol, unreacted water, and hydrogen chloride were removed by vacuum distillation, yielding 151 g of a liquid methoxy group-bearing 30 polyorganosiloxane.

The resulting polyorganosiloxane is represented by the compositional formula: $R^3_m Si(OR^5)_p(OH)_q O_{(4-m-p-q)/2}$ wherein m = 1.0, the organic substituent R³ attached to a Si atom through a Si-C bond consisted of 100 mol% phenyl, p = 1.5, R⁵ is methyl, and q = 0.2, containing 100 mol% of trifunctional siloxane units. It looked a

colorless clear liquid and had a number average degree of polymerization of 3 and a number average molecular weight of 500.

Polyorganosiloxane-epoxy resin copolymer (organopolysiloxane content 16 wt%):



Curing accelerator:

triphenyl phosphine by Hokko Chemical K.K.

Parting agent:

Carnauba wax by Nikko Fine Products K.K.

Silane coupling agent:

KBM403 (γ -glycidoxypropyltrimethoxysilane) by Shin-Etsu Chemical Co., Ltd.

Properties of epoxy resin compositions were measured by the following methods.

(1) Spiral Flow:

Measured by molding at 175°C and 70 kgf/cm² for a molding time of 90 seconds using a mold in accordance with EMMI standards.

(2) Hardness When Molded:

Using the method described in JIS-K6911, a rod measuring 10×4×100 mm was molded at 175°C and 70 kgf/cm² for a time of 90 seconds. The hardness when hot was measured with a Barcol Impressor.

(3) Moldability

A QFP package of 14×20×3.5 mm was molded. Using an ultrasonic flaw detector, internal voids were detected.

(4) Flame Retardance:

5 A 1/16 inch thick sheet was molded and the flame retardance of the sheet was rated in accordance with UL-94 test specifications.

(5) Moisture Resistance:

10 A silicon chip measuring 1×1 mm on which aluminum wiring had been formed was adhesively bonded to a 14-pin dual in-line package (DIP) frame (Alloy 42), and the aluminum electrodes on the chip surface were wire bonded to the lead frame using 30 µm gold wire. The epoxy resin composition was then molded over the chip at 175°C and 70 kgf/cm² for a time of 120 seconds, and post-cured at 180°C for 4 hours. Twenty packages thus obtained were left to stand for 500 hours at 140°C and 85% relative humidity while being subjected to a bias voltage of 5V DC. The number of packages in which aluminum corrosion arose was counted.

(6) Solder cracking resistance

20 A flat package of 14x20x2.7 mm was molded using an epoxy resin composition. It was postcured at 180°C for 4 hours, allowed to stand in a thermostat chamber at 85°C and RH 85% for 168 hours, and immersed in a solder bath at a temperature of 220°C for 30 seconds. The outside of the 25 package were observed to inspect cracks.

Examples 1-8 and Comparative Examples 1-4

Epoxy resin compositions for semiconductor encapsulation were prepared by uniformly melt mixing the 30 components shown in Tables 1 and 2 in a hot twin-roll mill, followed by cooling and grinding. The measured properties of the compositions are shown in Tables 3 and 4.

Table 1

Table 2

Ingredients (pbw)	Comp. Ex.1	Comp. Ex.2	Comp. Ex.3	Comp. Ex.4
Epoxy resin (a)	64.52	58.32	-	-
Epoxy resin (b)	-	-	52.71	52.71
Curing agent (c)	35.48	35.48	-	-
Curing agent (d)	-	-	47.29	47.29
Polyorganos iloxane A	-	-	-	-
Polyorganos iloxane B	-	-	-	10
Polyorganos iloxane C	-	-	-	-
Polyorganos iloxane D	-	-	-	-
Inorganic filler	500	500	500	500
Curing accelerator	1.2	1.2	1.2	1.2
KEMGARD 1261	100	-	-	-
Antimony trioxide	-	4	-	-
Brominated novolac type epoxy resin	-	6.2	-	-
Parting agent	3	3	3	3
Carbon black	2	2	2	2
Silane coupling agent	1	1	1	1

Table 3

	Ex.1	Ex.2	Ex.3	Ex.4	Ex.5	Ex.6	Ex.7	Ex.8
Spiral flow (cm)	95	85	83	80	65	90	115	80
Hardness when molded	80	75	78	81	75	80	73	70
Moldability	OK							
Flame retardance	V-0							
Moisture resistance	0/20	0/20	0/20	0/20	0/20	0/20	0/20	0/20
Solder cracking resistance	0/5	0/5	0/5	0/5	0/5	0/5	0/5	0/5

5 Table 4

	Comp. Ex.1	Comp. Ex.2	Comp. Ex.3	Comp. Ex.4
Spiral flow (cm)	98	90	125	70
Hardness when molded	80	80	78	65
Moldability	OK	OK	NG	NG
Flame retardance	burned	V-0	V-1	V-0
Moisture resistance	0/20	7/20	0/20	0/20
Solder cracking resistance	4/5	3/5	0/5	0/5

From Tables 3 and 4, it is evident that the semiconductor encapsulating epoxy resin compositions of the invention are able to cure into products which have excellent flame retardance and solder crack resistance. The compositions are effectively moldable and moisture resistant. The compositions have no ill effects on human health and the environment because they contain no halogenated epoxy resins or antimony trioxide.

Examples 9-16 and Comparative Examples 5-9

Epoxy resin compositions for semiconductor encapsulation were prepared by uniformly melt mixing the components shown in Tables 5 and 6 in a hot twin-roll mill.

followed by cooling and grinding. The measured properties of the compositions are shown in Tables 7 and 8.

Table 5

Ingredients (pbw)	Ex.9	Ex.10	Ex.11	Ex.12	Ex.13	Ex.14	Ex.15	Ex.16
Epoxy resin (a)	47.55	47.55	7.06	7.06	-	-	-	-
Epoxy resin (b)	-	-	-	-	46.41	6.89	57.55	17.03
Curing agent (c)	-	-	-	-	-	-	36.20	32.97
Curing agent (d)	46.20	46.20	42.94	42.94	47.34	43.11	-	-
Epoxy-modified silicone resin (Amount of siloxane added)	6.25	6.25	50.0	50.0	6.25	50.0	6.25	50.0
Inorganic filler	500	500	500	500	850	850	760	760
Curing accelerator	1.2	1.2	1.2	1.2	1.2	1.2	1.2	1.2
KEMGARD 1261	80	100	80	100	-	-	100	100
KEMGARD 911C	-	-	-	-	10	10	-	-
Antimony trioxide	-	-	-	-	-	-	-	-
Brominated novolac type epoxy resin	-	-	-	-	-	-	-	-
Parting agent	3	3	3	3	3	3	3	3
Carbon black	2	2	2	2	2	2	2	2
Silane coupling agent	1	1	1	1	1	1	1	1

Table 6

Ingredients (pbw)	Comp. Ex.5	Comp. Ex.6	Comp. Ex.7	Comp. Ex.8	Comp. Ex.9
Epoxy resin (a)	53.33	53.33	-	-	46.5
Epoxy resin (b)	-	-	-	-	-
Curing agent (c)	-	-	-	-	34.8
Curing agent (d)	46.67	46.67	37.5	37.5	-
Epoxy-modified silicone resin (Amount of siloxane added)	0 0	0 0	62.5 10	62.5 10	12.5 2
Inorganic filler	500	500	500	500	500
Curing accelerator	1.2	1.2	1.2	1.2	1.2
KEMGARD 1261	-	80	-	80	-
KEMGARD 911C	-	-	-	-	-
Antimony trioxide	-	-	-	-	4
Brominated novolac type epoxy resin	-	-	-	-	6.2
Parting agent	3	3	3	3	3
Carbon black	2	2	2	2	2
Silane coupling agent	1	1	1	1	1

5 Table 7

Table 8

	Comp. Ex.5	Comp. Ex.6	Comp. Ex.7	Comp. Ex.8	Comp. Ex.9
Spiral flow (cm)	125	100	55	40	110
Hardness when molded	80	78	82	78	80
Moldability	OK	OK	NG	NG	OK
Flame retardance	burned	V-1	V-1	V-0	V-0
Solder cracking resistance	5/5	5/5	1/5	2/5	0/5

From Tables 7 and 8, it is evident that the
5 semiconductor encapsulating epoxy resin compositions of the
invention are able to cure into products which have
excellent flame retardance. The compositions are
effectively moldable and moisture resistant. The
compositions have no ill effects on human health and the
10 environment because they contain no halogenated epoxy resins
or antimony trioxide.

Japanese Patent Application Nos. 11-118155 and 11-
118408 are incorporated herein by reference.

Although some preferred embodiments have been
15 described, many modifications and variations may be made
thereto in light of the above teachings. It is therefore to
be understood that the invention may be practiced otherwise
than as specifically described without departing from the
scope of the appended claims.